

Application of Isotope Dilution Mass Spectrometry for Reference Measurements of Cadmium, Copper, Mercury, Lead, Zinc and Methyl Mercury in Marine Sediment Sample

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Abstract. Marine sediment was selected as a test sample for the laboratory inter-comparison studies organized by the Environment Laboratories of the International Atomic Energy Agency. The analytical procedure to establish the reference values for the Cd, Cu, Hg, Methyl Hg, Pb and Zn amount contents was based on Isotope Dilution Inductively Coupled Plasma-Mass Spectrometry (ID ICP-MS) applied as a primary method of measurement. The Hg and Methyl Hg determination will be detailed more specifically because of the problems encountered with this element, including sample homogeneity issues, memory effects and possible matrix effects during the ICP-MS measurement stage. Reference values, traceable to the SI, with total uncertainties of less than 2% relative expanded uncertainty ($k=2$) were obtained for Cd, Cu, Zn and Pb and around 5% for Hg and CH_3Hg .

Key words: ID ICP-MS, cadmium, copper, lead, zinc, mercury, methyl mercury, sediment sample, uncertainty, traceability, validation

Introduction

The reference values of element amount contents in the marine coastal sediment sample were established by the application of measurement procedures based on Isotope Dilution Mass Spectrometry (IDMS) method. Because of its well understood measurement process the IDMS is recognised as primary method of analysis (according to the CCQM definition) [1]. The main advantage of using this technique is obtaining the SI traceable results with low combined uncertainty. The determinations of element amount contents were a part of the procedure used for the preparation of the IAEA candidate certified reference material. In this work five elements (Cd, Cu, Hg, Pb, Zn) present in the sediment at the very different levels (from 0.049 mg/kg Hg to 156 mg/kg Zn) were determined by IDMS.

Materials and Methods

Five certified isotopic reference materials were used in these measurements for the blends preparation (^{111}Cd from IRMM-622, ^{202}Hg from IRMM-640, ^{65}Cu from IRMM-632, ^{206}Pb from NIST-991, ^{68}Zn from

IRMM-654). The blends were prepared by weighing of the aliquots of the sample and the optimised masses of the isotopic certified reference materials (spikes). Then the blends were digested in the closed microwave system in the mixture of acids. Before the analyses samples were appropriately diluted or in case of cadmium the matrix separation was carried out. The isotopic ratios of cadmium, copper and zinc were measured by two types of inductively coupled plasma mass spectrometers: the quadrupole (Q-ICPMS, XSERIES, Thermo Scientific) and sector field high resolution (SF-HR-ICP-MS, ATTOM, NU Instruments). At the aim of minimizing the influence of spectral interferences on the results the Collision Cell Technology (CCT) or medium resolution were used. The medium resolution (MR-ICP-MS) was applied for copper and zinc isotopic ratio measurements and the Q-ICP-MS with CCT mode was used for Cd and Cu isotopic ratio analyses.

Results and Discussion

The IDMS measurement step was validated according to the requirements of ISO/IEC 17025. The validation of the measurement process encompassed: 1. Using isotopic

Table 1. Main contributors to the total uncertainty on the total copper mass fraction

Uncertainty contribution	SF-ICP-MS, MR	Q-ICP-MS, standard mode	Q-ICP-MS, CCT mode
Uncertainty on ^{65}Cu concentration in spike	60.4%	83.2%	85.1%
Correction for dead time effects	16.2%	1.2%	0.9%
Correction for mass discrimination effects	9.6%	2.7%	1.9%
Correction for blend ratio measurements	7.9%	4.8%	4.1%
Correction for moisture content	3.3%	4.5%	4.6%
Correction for sample weighing	1.9%	2.7%	2.7%
Others	0.7%	0.9%	0.7%

CRMs for blend preparation and for mass discrimination corrections. 2. Identification of the majority of potentially significant factors influencing the results (the isotopic equilibrium, ICP-MS parameters, the procedural blank, moisture content, memory effect, the isotopic ratio stability, dead time). 3. Mathematical modelling of the entire measurements process and demonstrating the traceability to the mole or kilogram SI units. 4.

The estimation of the combined uncertainty of the results was done according to the ISO/GUM guide. 5. Comparative studies of the different analytical procedures and applying two types of instruments (Q-ICP-MS, SF-HR-ICP-MS) for measurements. The adequate results obtained by applying different analytical approaches were identical within the range of the expended uncertainty. Estimated combined uncertainties for Cd, Cu, Pb and Zn were around 2%, only in case of mercury the expended uncertainty was 3.6%. The total uncertainty budget was calculated on the basis of the IDMS equation by using the method of propagation of

standard uncertainties according to ISO guidelines. The main uncertainty contributions to the total uncertainty in the case of Cu are presented in Table 1.

Conclusion

ID ICP-MS applied as a primary method of measurements for environment matrices leads to SI traceable values with small combined uncertainties. In conclusion, the obtained results have shown that three measurements approaches used in this studies have given identical values within the range of uncertainty.

References

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