

Microstructural, morphological and optical properties of sprayed thin films of indium doped gadolinium oxide ($Gd_2O_3: In$)

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Abstract. This paper featured a study on undoped and Indium doped Gadolinium oxide $Gd_2O_3: In$ thin films, elaborated on a glass substrates at temperature of 500 °C by homemade Spray Pyrolysis technique, at different Indium concentrations as follow 0, 2, 4, 6 and 8 at %. This thin layers, where a subjects to a numerous characterization techniques to study the effect caused by introducing the dopant element "Indium" in Gadolinium oxide lattice on the structural properties (X-Ray Diffraction and Raman spectroscopy) and optical properties. The structural characterization carried by the X-ray diffraction (XRD) reveals a polycrystalline Monoclinic B-type structure for all $Gd_2O_3: In$ thin films. Moreover, these findings are verified by the Raman spectroscopy results. Concerning the optical properties of our thin films, the optical measurements carried by UV-VIS-NIR spectrophotometer shows an increase in the transmittance value within the visible region [370-900 nm] and in the band gap energy value by raising Indium doping rate from 0 at % to 6 at %, also the disorder caused inside the thin films were estimated by the Urbach equation. That said, the 2 at % Indium doped gadolinium oxide thin film provides interesting results that can be applied in solar cells as an optical window material.

1 Introduction

Through the years, the transparent conducting oxides or TCO's were and still used in a variety of domains due to their electrical, optical, structural and morphological properties. The rare earth oxides more precisely Lanthanide oxides belongs to the TCO's family and gadolinium oxide ' Gd_2O_3 ' is one of them. The gadolinium oxide has a three-crystal structure variant as follow: Cubic ($c-Gd_2O_3$) [1] Monoclinic ($m-Gd_2O_3$) [2] and finally Hexagonal structure ($h-Gd_2O_3$) that present themselves through a variation of temperature [3] and pressure [4].

Gadolinium oxide is an n-type semiconductor, large band gap, high order of transparency and high dielectric constant allowing him to be used firstly as thin film in optoelectronic devices [5], alternative gate dielectric of SiO_2 [6], spectral converter [7]. Secondly as nanoparticles in biomedical domain as a contrast agent for imaging [8] either in its undoped form or by introducing some doping elements to Gd_2O_3 matrix such as Yb^{3+} [9] Eu^{3+} [10] Tb^{3+} [11] Bi^{3+} [12] toward investigating and refining the thin film properties to each desirable applications cited before.

Gadolinium oxide thin films are obtained through a variety of elaboration technics characterize either by their low cost: Sol-Gel method [13], Ultrasonic Spray Pyrolysis [8], Precipitation method [14] or expansive methods such as Molecular Beam Epitaxy [15], Pulsed Laser Deposition [2], Atomic Laser Deposition [16], Radio Frequency magnetron sputtering [17] and Metalorganic Chemical Vapor Deposition method [6].

Herein this study, we exhibit for the first time, to our knowledge, a study of the microstructural and optical

properties of the undoped and Indium doped Gadolinium oxide thin films acquired by a Spray Pyrolysis technic.

2 Experimental method

2.1 Material synthesis

Indium doped Gadolinium oxide, $Gd_2O_3: In$, thin films were synthesized by a homemade spray pyrolysis technique; a description of the technique was explained in [18]. Using a precursor prepared by dissolving Gadolinium Chloride Hexahydrate ($GdCl_3 \cdot 6H_2O$; purity 99 %) as matrix and Indium chloride ($InCl_3$) as Indium dopant source into a 30 ml of distilled water.

Different concentration of Indium were used 0 at, 2 at, 4 at, 6 at and 8 at % to study the physical properties of the above thin layers.

Each sample was obtained by fixing certain parameters such as, solution concentration of 0,02 M, deposition time to 6 min, substrate temperature to 500 °C, spraying rate to 2 ml/min.

Each sample was elaborated by transporting the desirable solution through tube with a flow rate of 2 ml/min in order to spray it to fine droplets by the help of a carrier gas presented by air to reach the clean, preheated glass substrates at 500 °C. The deposition parameters are summarized in Table 1.

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Table 1. Experimental parameters used for $Gd_{2-x}In_xO_3$ ($x=0, 2, 4, 6$ and 8 at %).

Deposition parameters	values
Precursor source	GdCl ₃ , 6H ₂ O
Concentration	0,02 M
Dopant source	InCl ₃ (0, 2, 4, 6 and 8 at %)
Temperature	500 °C
Deposition time	6 min
Nozzle-substrate distance	40 cm
Flow rate of solution	2 ml/min
Solution volume	30 ml
Substrate	Micro slide glass

2.2 Materials characterization

The structural properties of the undoped and doped thin films were investigated by using X-ray diffraction (XRD) system with Cu K α radiation ($\lambda = 1.5418 \text{ \AA}$), the X-ray generator's power was 1000W (40 kV, 25 mA), the scan range; from 7.5° to 70° , the scanning speed was 1 s/step, where one step corresponds to 0.029° . To reveal the vibrational modes between atoms in the lattice we have used a micro-Raman spectroscopy SENTERRA II, by fixing the source laser emitting at 532 nm. Furthermore, to study the optical properties, we have used SHIMADZU UV-Vis-NIR spectrophotometer, the wave number range was from 220 to 1600 nm.

3 Results and discussions

3.1 X-ray diffraction results

Figure 1 illustrates the X-ray diffraction (XRD) patterns of undoped and Indium doped Gadolinium oxide $Gd_{2-x}In_xO_3$ ($x=0, 2, 4, 6$ and 8 at %) thin layers prepared at 500°C .

The X-ray diffraction patterns in Figure 1 match with the monoclinic B-type structure (Space group C2/m) without any secondary phases of the source Gd_2O_3 according to Joint Committee on Powder Diffraction Standards (JCPDS, card number 01-076-7419 nor the Indium, dopant element, such as In_2O_3 or some Indium Gadolinium alloys or coupled were detected. Proving

that Indium ions were successfully incorporated into lattice sites of gadolinium oxide " Gd_2O_3 " and act as dopant.

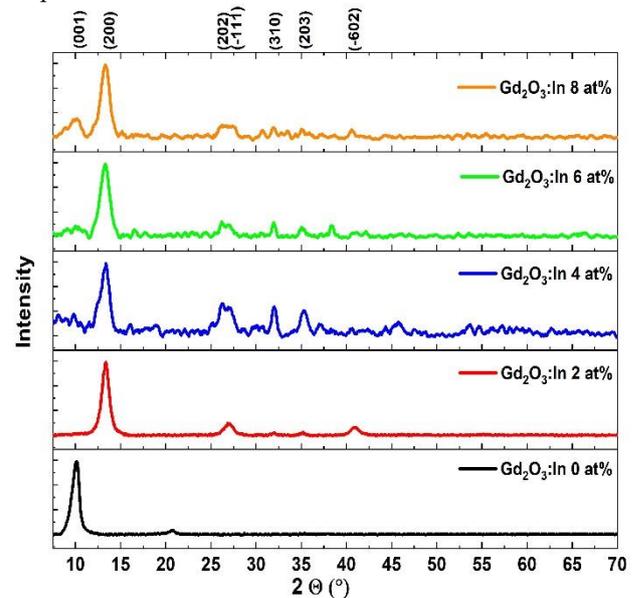


Fig. 1. X-ray diffraction of $Gd_{2-x}In_xO_3$ ($x=0, 2, 4, 6, 8$ at %) thin layers deposited on glass substrates.

As can be seen, all films in Figure 1 presents a polycrystalline texture with a preferred orientation that changes from (001) for the undoped to (200) for Indium doped Gadolinia thin layers plus those (200) peaks were shifted towards higher angles which it can be caused by the difference in the ionic radius of Gd^{3+} (0.938 \AA) and In^{3+} (0.80 \AA). Furthermore, the intensity of the privileged direction (200) decreases with the increase of Indium concentration in the thin films, this reduction can be as a result of the Indium atoms interfering to diminish the nucleation center of Gd_2O_3 [19]. Moreover, the highest intensity of this peak is attached to the 2 at % In leading to a better crystallinity for our Gd_2O_3 thin layer. Also, the increase of the doping rate of Indium starting from 4 at% to 8 at%, we notice that X-ray patterns are a little bit rough compared to the 2 at% or undoped films.

The lattice constants, a; b; c, were calculated using peak's positioning through monoclinic structure formula:

$$\frac{1}{d_{hkl}^2} = \frac{1}{\sin^2(\beta)} \cdot \left(\frac{h^2}{a^2} + \frac{\sin^2(\beta) \cdot k^2}{b^2} + \frac{l^2}{c^2} - \frac{2 \cdot h \cdot l \cdot \cos(\beta)}{ac} \right) \quad (1)$$

Where d_{hkl} the inter-planar distance and h, k, l are the miller indices and $\beta = 100.182^\circ$ which is the angle between the lattice parameter a and b and c successively.

The average crystallites size (D) was calculated using the Debye-Scherrer formula [2]:

$$D = \frac{0.9 \cdot \lambda}{\beta \cdot \cos(\theta)} \quad (2)$$

The micro strain (ϵ) was calculated by the following equation:

$$\epsilon = \frac{\cos(\theta) \cdot \beta}{4} \quad (3)$$

The micro strain (ϵ) was calculated Williamson-Hall equation [19]

$$\beta \cos(\theta) = \varepsilon(4\sin(\theta)) + \frac{0.9\lambda}{D} \quad (4)$$

The dislocation density (δ) is a parameter that exhibits crystal lattice discontinuity, it was calculated by the equation:

Table 2. Structural and lattice parameters and micro strain of In doped Gd₂O₃ thin films

Gd ₂ O ₃ : In	d ₂₀₀ (Å)	2 θ (°) (200)	FWHM	ε 10 ⁻³	ε 10 ⁻³ (W-H)	δ 10 ⁻² (nm ⁻²)	a (Å) “(200)”	b (Å) “(310)”	c (Å) “(203)”	D (nm)
Undoped	-	-	1.0965	4.7661	2.75	1.890	14.129 /(-201)	3.57	8.848	7.2726
2 at %	6.6071	13.328	1.15851	5.0207	-9.23	2.098	13.487	3.627	8.95	6.9038
4 at %	6.7118	13.286	1.23735	5.3627	-13.91	2.394	13.423	3.58	7.842	6.463
6 at %	6.6386	13.259	1.27135	5.5102	-13.88	2.527	13.557	3.6	7.855	6.2906
8 at %	6.6119	13.257	1.28181	5.5555	-13.22	2.568	13.558	3.79	7.814	6.2392

$$\delta = \frac{1}{D^2} \quad (5)$$

Where λ the wavelength of the incident X-Ray beam of Cu K α 1.54056 Å, β full width at the half maximum (FWHM) of the diffraction line and θ is the position of the diffraction peak considered.

The microstructural parameters of prepared undoped and indium doped gadolinium oxide “Gd₂O₃” thin layers are illustrated in Table 2.

Firstly, starting with lattice parameters of the monoclinic structure. The JCPDS card number 01-076-7419 presents them as follow a= 14.195 Å, b=3.566 Å and c= 8.77 Å [9] Through calculations of lattice parameters for our undoped and Indium doped Gd₂O₃ thin films, using Equation 1 then presenting the results obtained in Table 2.

Using the monoclinic structure JCPDS card mentioned above as a footing for comparison with lattice parameters calculated and presented in table 2. We notice, on one hand, that the undoped Gd₂O₃ thin film its lattice parameters (a, b, c) are slightly different from that of the standard Gd₂O₃ monoclinic crystals indicating that our layer of Gadolinia has a very good crystalline structure. On the other hand, the lattice parameters for In doped Gd₂O₃ films are inferior to the standard Gadolinia crystals. This reduction can be relate to a substitution of Gd³⁺ ions having an ionic radius of (0.938 Å) with a much smaller one presented in In³⁺ (0.80 Å).

Secondly, moving to crystallite size (D) parameter of the Gd_{2-x}In_xO₃ (x= 0, 2, 4, 6 and 8 at %) thin films. This parameter (D) is calculated using Eq. 2 and presented in Table 2. By analyzing the results obtained (Tab. 2), we reached a conclusion of that by increasing In dopant concentration the mean crystallite size decreases from 7.2726 nm for undoped Gd₂O₃: In layer to 6.2392 nm for Gd_{2-0.08}In_{0.08}O₃ (8 at %) thin layer (Fig. 3). Where Gd_{2-0.02}In_{0.02}O₃ (2 at %) have the largest grain size between the doped samples and exhibits a better crystallinity compared to the others. So, the decrease of the mean crystallite size of our layers can be explain by the difference in the size of the radius between Gd³⁺ ions (0.938 Å) and In³⁺ ions (0.80 Å) where it may have caused

a stress “ε”, leading to a disturbance in the crystallite growth kinetics [12].

Finally, by using Eq 3 to determine the strain parameter “ε” value in our thin layers, we had an observation of an increase in the stain “ε” as a function of

increasing the In doping concentration as shown in Table 2. This act causes a peak broadening “FWHM” (Table 2) affecting the average crystallite size to decrease (Fig. 2).

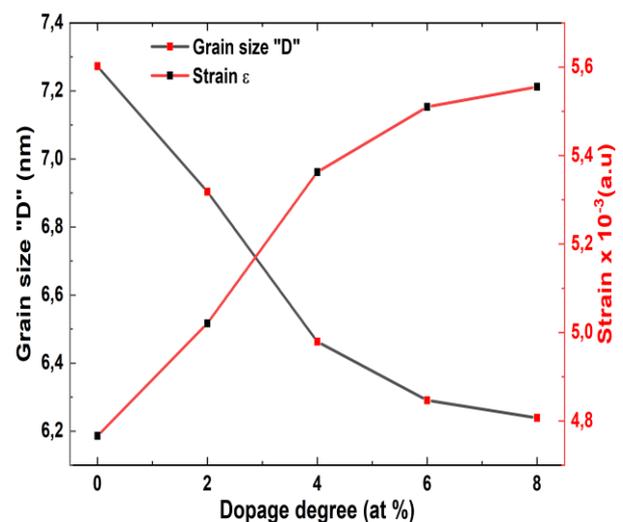


Fig. 2. Average grain size, micro strain as a function of Undoped and In doped Gd₂O₃ thin films.

So, to further more understand the nature of the micro strain Eq 4 was used. The Williamson-Hall equation is used to determine the nature and strain value residue into Gd_{2-x}In_xO₃ lattice by determining the slope of ($\beta \cos(\theta)$) as a function of ($4\sin(\theta)$) of different peaks [19]. We’ve obtained a negative values corresponding to In doped Gd₂O₃ layers indicating the presence of a compressive strain in the Gd_{2-x}In_xO₃ films caused by the difference in the radius size (Fig. 3), acknowledging the substitution of Gd³⁺ ions (0.938 Å) with In³⁺ (0.80 Å) into the lattice. Moreover, we can see that the smallest value of the dislocation density “δ”, obtained through Eq 5, correspond to 2 at %, with a value of $\delta = 2.098 \cdot 10^{-2} \text{ nm}^{-2}$ (Tab. 2), indicating an occurrence of a minimal disturbance to Gd₂O₃ lattice.

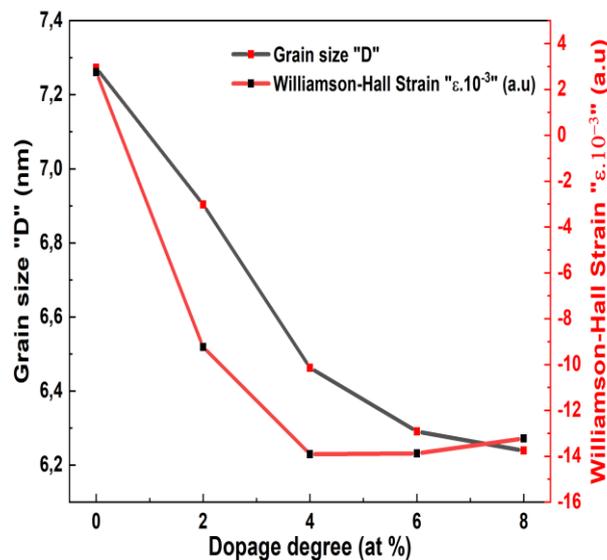


Fig. 3. Average grain size, Williamson-Hall micro strain as a function of Undoped and In doped Gd₂O₃ thin films.

3.2 Raman spectroscopy results

Raman spectroscopy is a powerful in-situ, nondestructive tool used to investigate the inelastic scattering of light emitted by the target, chemical bonds of the sample, through an analysis to identify the various Raman active vibrational, phonon, modes.

At room temperature, The-Raman experiment was carried out by using $\lambda=532$ nm line of 20 mW powered laser, where the laser beam was directly focused on undoped and In doped Gd₂O₃ samples through a microscope objective of (x50). The studied Raman Spectrum of Gd_{2-x}In_xO₃ (x= 0, 2, 4, 6 and 8 at %) are illustrated in Figure 4 and the obtained frequency are presented in the Table 3.

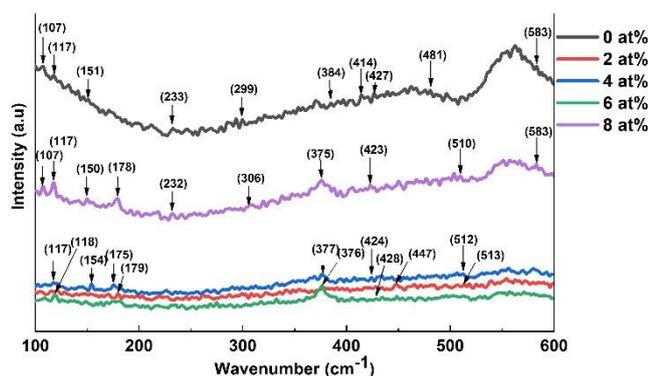


Fig. 4. Raman spectra of Undoped and In doped Gd₂O₃ thin films.

Based on the data presented in Table 3 of the micro-Raman wavenumber (cm⁻¹) analysis of the In doped Gd₂O₃ samples, we can confirm that the results obtained belongs to the monoclinic B-type structure (space group C 2/m) affirming the deduction, Gd₂O₃: In structure, obtained through the X-ray diffraction analysis above. Based on the factor group analysis, Zarembowitch et al [20] defined the Raman active modes in the monoclinic

Gd₂O₃ to be 21 Raman active modes at the center of Brillouine zone. In addition, Paul et al [21] have predicted those modes to be 14 related to A_g modes and 7 of them related to B_g modes.

Hence, In accordance with other articles [20,21,2] , Table 3 and Figure 4, we have identified in the interval of 100 – 600 cm⁻¹ , 9 Raman peaks plus a shoulder at 583 cm⁻¹for the undoped Gd₂O₃ layer then a 4, 6, 3 and a 9 Raman peaks for Gd₂O₃: In (2, 4, 6 and 8 at %) samples successively. In addition to that, there was an absence of any Raman peaks matching with In₂O₃ or any other unwanted impurity phase in our In doped Gd₂O₃ samples [22]. Furthermore, we noticed a certain shifting of the Raman peaks in our samples, which can be caused by In³⁺ ions introduced into Monoclinic Gd₂O₃ lattice, where this substitution leads to a change in the chemical band length of our molecule [23]. In one hand, based on Table 3, there was a slight shift of 3 to 5 cm⁻¹ towards the lowest wavenumbers (cm⁻¹) concluding the presences of micro strain in Gd₂O₃: In lattice [2]. On the other hand, we noticed a similar shift of the peaks towards the highest wavenumbers (cm⁻¹) generated by the reduction of the mean grain size in Gd₂O₃: In thin layers [21]. Finally, those results confirms our results and findings in X-ray diffraction paragraph

Table 3. Obtained data of Raman active modes and assigned Raman active modes of In doped Gd₂O₃

Gd ₂ O ₃ : In Obtained					Mode assignment factor group [20]	Reported wave-number (cm ⁻¹)
0 at%	2 at%	4 at%	6 at%	8 at%		
107	-	-	-	107	Ag	109
117	-	117	118	117	Bg	115
151	-	154	-	150	Ag	150
-	179	175	-	178	Ag	176
233	-	-	-	232		233[13]
299	-	-	-	306	Bg	299
384	376	377	376	375	Ag	385
414	-	-	-	-	Bg	416
427	-	424	428	423	Bg	427
-	447	-	-	-	Ag	442
481	-	-	-	-	Ag	483
-	513	512	-	510		514[13]
583	-	-	-	583	Ag	580 sh

3.3 Optical and optoelectronic properties of Gd₂O₃:

In thin films

The transmittance spectra of the undoped and In doped Gd₂O₃ (2, 4, 6 and 8 at %) thin films, obtained through the UV-Vis-NIR spectrophotometer in a spectra range of 220 to 1600 nm, are presented in Figure 5.

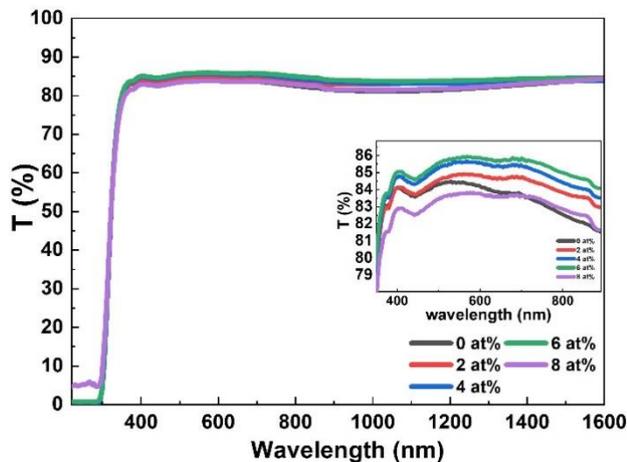


Fig. 5. Transmittance graph of the Gd_{2-x}In_xO₃ (x= 0, 2, 4, 6, 8 at) thin films. Insert; display a zoom in the visible region of the elaborated Gd₂O₃:In thin films.

These illustrations (Fig 5) are done after the subtraction of the transmission of the glass substrates.

To begin, the transmittance value of the undoped gadolinium oxide reaches a round value of 83,46 % in the visible spectra range, 370 -900 nm (figure 5), which agrees with previous research results that have been done placing the transmittance of gadolinium oxide thin film in a range between 70 – 90 % [2]. Where the variance of the value is dependable on the technics and parameters of elaboration used to obtain the thin films [2,6].

Moreover, slight incrementation in transmittance value was observed as a function of Indium doping in gadolinium oxide, leading to a maxima of T =85.243 % for the 6 at % indium doped. Also, refractive index, n, was calculated using the following equation and presenting it in figure 6 [24]:

$$n = \frac{1-R}{1+R} + \left(\frac{4R}{(1-R)^2} - k^2 \right)^{\frac{1}{2}} \quad (6)$$

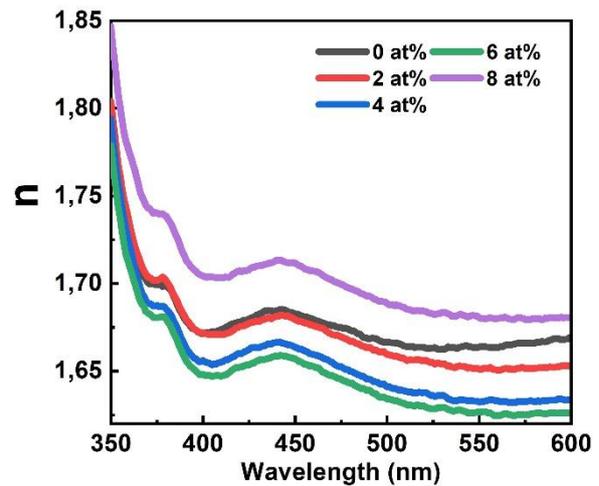


Fig. 6. Refractive index “n” of Undoped and In doped Gd₂O₃ thin films.

Where R is the reflection and k is the extinction index.

Therefore, the values obtained for our thin films for a wavelength equals to 500 nm are between 1,666 for undoped gadolinium oxide and 1,688 for the Indium ratio of 8 at % (Table 4). The result obtained for the undoped gadolinium oxide thin film which is 1,666 is relatable to other research works such as M. Misha et al. [2] that found it in between 1,6 and 1,8 for different temperatures and 1,6 and 1,7 for different oxygen partial pressure, also Q. S. Johnson et al.[10]

In addition, the graphs obtained in figure 7 shows an increase in the absorbance from undoped gadolinium oxide to 2 at then 4 at % indium doped gadolinium oxide respectively. This increase can be correlated with crystallite average size where the diminution amplify the light interactions with our thin films leading to an augmentation in it [25].

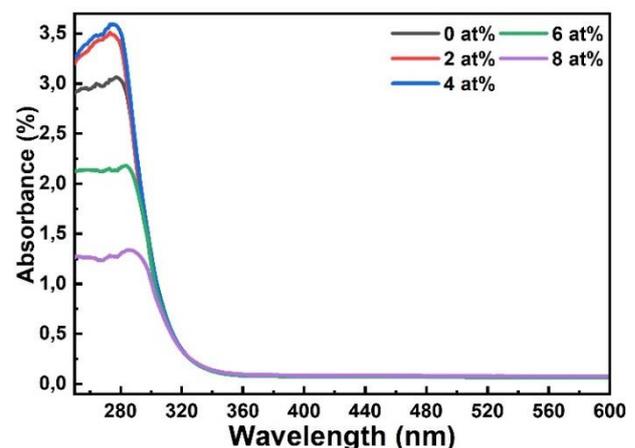


Fig. 7. Absorbance spectra of the Gd_{2-x}In_xO₃ (x= 0, 2, 4, 6, 8 at %) thin layers deposited on glass.

The band gap energy E_g was estimated through Tauc model relating the photon energy hv and the absorption coefficient α [6]:

$$ahv = C (hv - E_g)^r \quad (7)$$

Where C is a constant, E_g is the gap energy « eV », hu is phonon energy « eV » and « r = 1/2 » for direct band gap. By scanning the entire energy domain The resolution

of this equation is done graphically by plotting $(\alpha h\nu)^2$ as a function of the energy of a photon $h\nu$ (eV) and extrapolating the linear part of α up to the axis of the abscissa ($\alpha = 0$) and finally obtaining E_g value [24]. So, the band gap energies of our $Gd_{2-x}In_xO_3$ ($x = 0, 2, 4, 6$ and 8 at %) thin films were plotted and presented in figure 8.

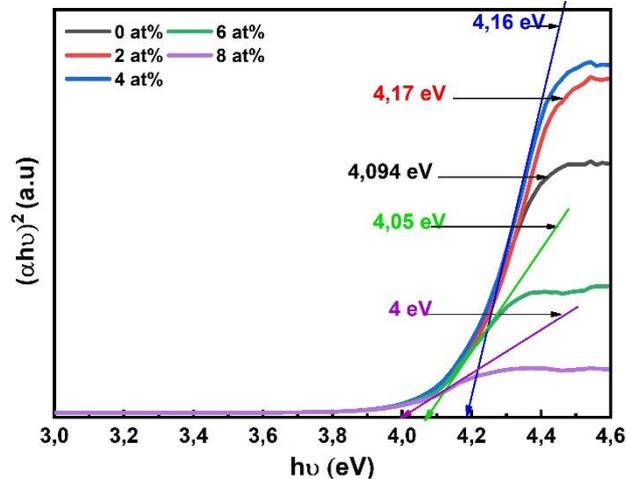


Fig. 8. Gap energy plotted by $(\alpha h\nu)^2$ vs $h\nu$ of the Undoped and In doped Gd_2O_3 thin films.

The use of Tauc plot pointed out, the elaborated thin films have a direct band gap, which is in accordance with various reported results [2]. Hence, the band gap value of the elaborated undoped gadolinium oxide thin film is $E_g = 4,09$ eV, this results is similar to the one obtained by M. Pattabi et al. [17], we note also that band gap value of gadolinium oxide with a monoclinic structure obtained through a DFT calculation is $E_g = 3,8$ eV [26]. So, after introducing the dopant element, Indium, at a different concentration rate we notice a blue shift in the band gap value of Indium 2 and 4 at % at 4,17 eV and 4,16 eV respectively (Fig 8). This increase is related to the augmentation of the carrier electrons introduced, leading to a shift and merge of Fermi level into the covalent band, which is known as Burstein-Moss effect [27]. Concerning the 6 and 8 at % of Indium doped gadolinium oxide thin films, a red shift is occurring (Fig 8) caused by an emergence of localized energy states within the band gap [28].

Furthermore, to investigate more the absorbance data, we estimated the disorder by Urbach tail using the following equation [18]:

$$\alpha = \alpha_0 e^{\left(\frac{h\nu}{E_u}\right)} \quad (8)$$

Where α_0 is a constant, E_u is the Urbach energy that highlights the intrinsic anomalies of lattice presented in structural disorder and defectiveness, it's determined by the slope of the exponential edge calculated by fitting linear portions of $\ln(\alpha)$ in terms of the photon energy $h\nu$ then represented in Figure 9. We found that the Urbach energy increases as a function of Indium doping concentration, where it's increases from $E_u = 0,205$ eV for the undoped gadolinium oxide thin film to $E_u = 0,239$ eV for Indium ratio of 8 at %. Based on the work of A.F. Zatsepin et al. [29] the increase in the degree of disorder is related with particle size effect, where grains size

reduction causes a thrivingness of defectiveness (Grain size, Tab. 2).

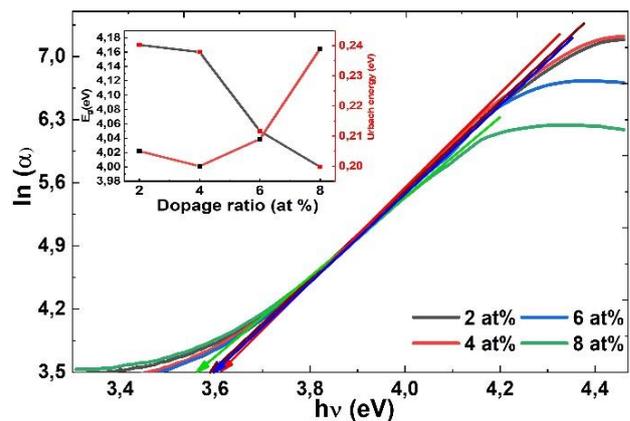


Fig. 9. $\ln(\alpha)$ vs photon energy (eV) spectra of the $Gd_{2-x}In_xO_3$ ($x = 2, 4, 6, 8$ at %) thin layers. The insert, the band tail width, E_u upon the $Gd_{2-x}In_xO_3$ ($x = 2, 4, 6, 8$ at %) thin films.

Table 4. Optical parameters of $Gd_{2-x}In_xO_3$ ($x = 0, 2, 4, 6$ and 8 at %) thin films

Gd ₂ O ₃ : In	Average T (%) in the [370-900 nm]	E _g (eV)	E _u (eV)	Refractive index, n
Undoped	83,46	4,09	0,205	1,666
2 at %	84,208	4,17	0,205	1,659
4 at %	84,856	4,16	0,200	1,641
6 at %	85,243	4,05	0,209	1,634
8 at %	83,11	4	0,239	1,688

4 Conclusion

To conclude, a good quality of undoped and Indium doped Gd_2O_3 (0, 2, 4, 6 and 8 at %) thin layers have been elaborated on a glass substrate using a homemade Spray pyrolysis technic. A number of properties were investigated using both characterization technics and semiempirical equations to determine the structural, morphological and optical properties.

The structural properties show a well crystallize Monoclinic B-type structure for all the thin films elaborated Gd_2O_3 : In (0, 2, 4, 6 and 8 at %) with a change in the preferential orientation from (0 0 1) for the undoped Gd_2O_3 to (0 0 2) orientation for the In doped Gd_2O_3 layers. Furthermore, the Raman spectroscopy support the X-ray diffraction results obtained. Concerning the optical properties, Indium ratio of 2 at % presents a good transmittance in the visible range (84,208 %) with a band gap equals to 4,17 eV allowing him to absorbance a certain short wave in the UV and also a good refractive index (1,659). This improvement of parameters may push it to be used as an optical window for solar cells.

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