Treatment of Waterborne Coating Wastewater by Coagulation Sedimentation Fenton Oxidation Process

Chenxi Yang 1,2,3,4,*, Jian Wang 1,2,3,4

1 Institute of Land Engineering and Technology, Shaanxi Provincial Land Engineering Construction Group Co., Ltd., Xi'an 710075, China
2 ShaanXi Provincial Land Engineering Construction Group Co., Ltd., Xi'an 710075, China
3 Key Laboratory of Degraded and Unused Land Consolidation Engineering, the Ministry of Land and Resources, Xi'an 710075, China
4 Shaanxi Provincial Land Consolidation Engineering Technology Research Center, Xi'an 710075, China

Abstract: waterborne coatings have become the development trend of the coating industry because of their less use of organic solvents and good environmental protection performance. However, the wastewater produced in the production process of waterborne coatings has become a difficult pollution source. Based on this, this paper uses coagulation sedimentation method and Fenton oxidation method to treat waterborne coating wastewater. Taking the COD of treated wastewater as the research index, this paper studies various factors affecting the treatment effect of coagulation sedimentation and Fenton oxidation. Through single factor analysis, the results show that the best dosage of PFS and PAM is PFS: 8 g/L and PAM: 0.8 g/L; In Fenton oxidation, the dosage of H2O2 is 9.98mL/L, the dosage of FeSO4 ⋅ 7H2O is n (Fe2+): n (H2O2) = 1:6, pH is 2, and the dosage times are 2.

Key words: coagulation sedimentation, Fenton oxidation, waterborne coating wastewater

1. Introduction

The water-based coating takes water as the dispersion medium. During the construction and subsequent drying process, the water volatilizes and returns to nature in a gas state, which will not harm the human body and the environment. Therefore, with the enhancement of people's awareness of environmental protection and the government's efforts to improve environmental protection, the water-based coating has become more and more popular [1,2]. However, in the production process, water-based coatings will produce a large amount of wastewater with high chroma, turbidity and COD content [3,4]. Therefore, how to treat water-based coating wastewater in an economical and simple way to make it meet the national discharge standard has become a major problem in water pollution control.

The main components of waterborne coatings include waterborne acrylic resin (film-forming material), deionized water, film-forming additives, wetting agent, defoamer, leveling agent, pH regulator, preservative, thickener, filler, pigment, etc [5]. The main sources of wastewater are batching tank washing liquid and cleaning wastewater. In principle, the washing solution of batching tank is the wastewater after the dilution of the original coating sample solution, so its chroma, turbidity and COD content are large. At present, aiming at the basically known pollution sources and types of water-based coating wastewater, the methods for treating this kind of organic wastewater at home and abroad mainly include incineration method, coagulation sedimentation method, chemical oxidation method, electrolysis method, biological method and combined method [6].

Taking the waterborne coating wastewater produced by a waterborne coating production company in Guangdong as the research object, this paper discusses the treatment conditions of waterborne coating wastewater by coagulation sedimentation method and Fenton oxidation method. Taking the COD removal rate as the screening index, various factors were compared and analyzed and the optimization conditions were studied.

2. Experiment

2.1 Materials and instruments

The reflux condenser, 250 mL conical flask, heating jacket, acid burette and thermometer are all purchased from the local chemical glass station. Silver sulfate (Ag2SO4, AR), mercury sulfate (HgSO4, AR), sulfuric acid (H2SO4, AR), phenanthroline (AR), potassium hydrogen phthalate (AR), polymeric ferric sulfate (PFS, AR), polypropylamide (PAM, AR), H2O2 (30%), ferric sulfate heptahydrate (FeSO4⋅7H2O, AR) and sodium hydroxide (NaOH, AR) were purchased from Sinopharm group.
2.2 Coagulation sedimentation method
PFS is selected as coagulant and PAM as coagulant aid. Take several 100 mL water samples into the beaker, add 8 g/L PFS and 0.6 g/L PAM respectively for coagulation test, and conduct the experiment by selecting the stirring strategy of fast stirring for 1 min and slow stirring for 1.5 min, and then set the sedimentation time as 1h.

2.3 Fenton oxidation method
H₂O₂ dosage, FeSO₄·7H₂O dosage, pH value and dosing times are selected as the control conditions. According to the literature, the ratio of n (Fe²⁺): n (H₂O₂) is 1:5 ~ 8. Therefore, the amount of FeSO₄·7H₂O is calculated by calculating the amount of H₂O₂. The single factor test is carried out by controlling the variable method, so as to determine the optimal conditions of H₂O₂ dosage, FeSO₄·7H₂O dosage, pH value and dosing times.

3. Result Analysis

3.1 Treatment effect of coagulation sedimentation
Coagulation sedimentation can make the colloids and suspended solids in wastewater agglomerate into flocs through coagulation through the addition of coagulant, so as to precipitate and separate pollutants through sedimentation. The dosage of coagulant PFS and coagulant aid PAM determines the treatment efficiency of coagulation sedimentation. In the test, different amounts of PFS and PAM are added to treat coating wastewater, and the COD content of effluent is shown in Figure 1. With the increase of PFS dosage, the effluent COD decreases sharply. When the dosage increases to 8 g/L, the effluent COD removal rate reaches the peak. If the dosage of PFS continues to increase, the effluent COD decreases. The addition of PFS will hydrolyze and polymerize to form a variety of colloidal complex ions, which will promote the suspended solids and other pollutants in the coating to gather into flocs, so that the pollution can precipitate in the form of coagulation and sedimentation. However, when PFS is excessive, the opposite charges in the colloid will attract each other to form a new stable system, which reduces the coagulation and precipitation effect. Therefore, the optimal mass concentration of PFS is 8 g/L.

When the amount of PFS is controlled to be 8g / L, the effect of different coagulant aid dosage on COD removal rate is shown in Fig. 2 (b). The COD content of effluent decreased with the increase of PAM dosage, and reached the highest value at 0.8 g/L. When the dosage of PAM continues to increase, the COD of effluent will increase. This is because too many polymers will cover the surface of colloidal particles, and the polymers will repel each other due to the same charge, so that the colloidal particles cannot gather. Moreover, because PAM is a high molecular organic matter, when the dosage of PAM is excessive, only one part will precipitate through bridging adsorption, and the other part will remain in the supernatant to increase the cod. The removal rate becomes smaller. Therefore, the best mass concentration of PAM is 0.8 g/L.

![Figure 1](image1.png)

Figure 1. (a) effect of PFS dosage on effluent COD; (b) Effect of PAM dosage on effluent COD.

3.2 Treatment of coating wastewater by Fenton oxidation

2.2.1 Influence of H₂O₂ Dosage on Treatment Effect
Fenton process refers to the reaction between H₂O₂ and Fe²⁺ to produce ·OH. This hydroxyl radical can oxidize the organic matter in coating wastewater, so as to shorten the molecular chain segment of organic matter or mineralize it into CO₂ and H₂O. The ·OH produced by H₂O₂ is the core of chemical oxidation method. Therefore, firstly, the effect of H₂O₂ dosage on Fenton reaction is studied. Experimental conditions: the wastewater used in Fenton oxidation treatment is the wastewater treated under the best conditions of coagulation and sedimentation. At this time, the COD is 1380 mg / L. the effect of H₂O₂ on Fenton reaction is determined by fixing the dosage of FeSO₄·7H₂O as n (Fe²⁺): n (H₂O₂) = 1:6 And pH 2. The amount of H₂O₂ added is 1.66 mL/L, 3.33 mL/L, 6.65 mL/L, 9.98 mL/L, 13.30 mL/L and 16.63 mL/L respectively. After adjusting the pH to 2, detect the CODcr concentration and turbidity. The results are shown in Figure 2.

![Figure 2](image2.png)

Figure 2. Experimental results of CODcr and turbidity of different H₂O₂ dosages.
It can be seen from Figure 2 that the dosage of H$_2$O$_2$ will affect the treatment effect of wastewater. When the dosage was less than 9.98 mL/L, the CODcr removal rate increased significantly with the increase of H$_2$O$_2$ dosage; After 9.98 mL/L, the change of CODcr removal rate tends to be gentle. This is because when the H$_2$O$_2$ dosage is low, with the increase of H$_2$O$_2$ dosage, H$_2$O$_2$ will produce more ·OH, so as to oxidize organic matter, reduce COD and increase COD removal rate. When the dosage of H$_2$O$_2$ is greater than 9.98ml/l, even adding more H$_2$O$_2$ will not reduce the COD of wastewater, because adding more H$_2$O$_2$ will only produce ineffective decomposition and cannot increase the COD removal rate. Therefore, it can be seen from the above data that the optimal H$_2$O$_2$ dosage for wastewater treatment is 9.98 mL/L.

2.2.2 Effect of pH on Treatment Effect

According to the wastewater treatment mechanism of Fenton oxidation reaction, pH value will affect the direction of reaction. From the reaction formula Fe$^{2+}$ + H$_2$O$_2$ → Fe$^{3+}$ + OH + ·OH, it can be seen that Oh will be continuously consumed under acidic conditions, which promotes the reaction to the right. When the pH is low, it will inhibit the progress of the above reaction and reduce the production of ·OH. In order to explore the optimal pH value of oxidation treatment of waterborne coating wastewater, the optimal pH value of Fenton reaction in waterborne coating wastewater was studied by controlling single factor experiment. Experimental conditions: the experimental raw water is the wastewater treated under the best conditions of coagulation and sedimentation. The COD is 1380 mg/L, the dosage of H$_2$O$_2$ is 9.98mL/L, and the dosage of FeSO$_4$ · 7H$_2$O is 3.03g/L calculated according to n (Fe$^{2+}$): n (H$_2$O$_2$) = 1:6. The effect of pH on Fenton reaction was studied by changing pH, and the results are shown in Fig. 3.

4. Conclusion

The optimal treatment conditions were explored by coagulation sedimentation and Fenton oxidation. The following conclusions are drawn through the single factor test method:

(1) The effect of PFS and PAM dosage on effluent COD in coagulation sedimentation method is discussed through single factor experiment. The optimal conditions are PFS: 8 g/L and PAM: 0.8 g/L.

(2) According to the experimental results of Fenton oxidation reaction, the optimal reaction conditions for COD removal rate are determined: the dosage of H2O2 is 9.98mL/L; The dosage of FeSO$_4$ · 7H$_2$O is: n (Fe$^{2+}$): n (H$_2$O$_2$) = 1:6; PH 2; The number of dosing is 2.

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Reference


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